

# MATERIAL SAFETY DATA SHEET

**SRM Supplier:** National Institute of Standards and Technology  
Standard Reference Materials Program  
Bldg. 202 Rm. 211

Gaithersburg, Maryland 20899

**SRM Number:** 4334G  
**MSDS Number:** 4334G  
**SRM Name:** Plutonium-242  
Radioactivity Standard  
**Date of Issue:** 29 September 1999

**MSDS Coordinator:** Joylene W.L. Thomas  
**Phone:** (301) 975-6776 **FAX:** (301) 926-4751

**e-mail:** SRMMSDS@nist.gov  
**ChemTrec:** 1-800-424-9300

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## SECTION I. MATERIAL IDENTIFICATION

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**Material Name:** Plutonium-242 Radioactivity Standard

**Description:** SRM 4334G consists of radioactive plutonium-242 nitrate and nitric acid dissolved in 5 mL of distilled water. The resulting solution is 19 wt. % nitric acid.

**Other Designations:** Plutonium-242 in Nitric Acid (aqua fortis; hydrogen nitrate; azotic acid; engravers acid) Solution

| Name        | Chemical Formula | CAS Registration Number |
|-------------|------------------|-------------------------|
| Nitric Acid | HNO <sub>3</sub> | 7697-37-2               |

**DOT Classification:** Nitric Acid, UN2031

**Manufacturer/Supplier:** Available from a number of suppliers

**SRM 4334G is a radioactive material with a massic activity of approximately 30 Bq g<sup>-1</sup>. The hazard information supplied in this MSDS is for the Chemical Hazard Only! For the hazard documentation concerning the radioactive material, refer to the packaging information and insert sheet.**

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## SECTION II. HAZARDOUS INGREDIENTS

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| Hazardous Components | Nominal Concentration (%) | Exposure Limits and Toxicity Data             |
|----------------------|---------------------------|---|
| Nitric Acid          | 19                        | ACGIH TLV-TWA: 2 mg/kg or 5 mg/m <sup>3</sup> |
|                      |                           | OSHA TLV-TWA: 2 mg/kg or 5 mg/m <sup>3</sup>  |
|                      |                           | Human, Oral: LD <sub>50</sub> : 430 mg/kg     |
| Plutonium-242        | 2 x 10 <sup>-5</sup>      | N/A*  |

\*For the radiation hazard, refer to the packaging information and insert sheet.

### SECTION III. PHYSICAL/CHEMICAL CHARACTERISTICS

| Nitric Acid   |
|---|
| <b>Appearance and Odor:</b> A white to slightly yellow liquid that darkens to a brownish color upon aging and exposure to light; a strong, pungent odor |
| <b>Relative Molecular Mass:</b> 63.02   |
| <b>Density:</b> 1.107 (19 wt. % nitric acid)  |
| <b>Solubility in Water:</b> Soluble   |
| <b>Solvent Solubility:</b> Decomposes in alcohol  |

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#### SECTION IV. FIRE AND EXPLOSION HAZARD DATA

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**Autoignition Temperature:** N/A

|   |               |                     |     |
|---|---------------|---------------------|-----|
| <b>Flash Point:</b>                           | N/A           | <b>Method Used:</b> | N/A |
| <b>Flammability Limits in Air (Volume %):</b> | <b>UPPER:</b> | N/A                 |     |
|   | <b>LOWER:</b> | N/A                 |     |

**Unusual Fire and Explosion Hazards:** Although nitric acid does not burn, it is a powerful oxidizing agent that can react with combustible materials to cause fires.

**Extinguishing Media:** Use extinguishing media that is appropriate to the surrounding fire. Use a water spray to dilute nitric acid and to absorb liberated oxides of nitrogen.

**Special Fire Procedures:** Fire fighters should wear a self-contained breathing apparatus (SCBA) with a full face piece in the pressure demand or positive mode and other protective clothing.

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## SECTION V. REACTIVITY DATA

**Stability:**            **X**    **Stable**                          **Unstable**

**Conditions to Avoid:** Avoid contact with combustible and other incompatible materials.

**Incompatibility (Materials to Avoid):** Keep nitric acid away from organic materials, plastics, rubber, and some forms of coatings. Nitric acid is incompatible with chlorine and metal ferrocyanide.

See Section IV: *Unusual Fire and Explosion Hazards.*

**Hazardous Decomposition or Byproducts:** Hazardous decomposition of nitric acid can produce various nitrogen oxides, including nitric oxide (NO), nitrogen dioxide (NO<sub>2</sub>), nitrous oxide (N<sub>2</sub>O), as well as nitric acid mist or vapor.

**Hazardous Polymerization:** \_\_\_\_\_ Will Occur                      **X** Will Not Occur

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## SECTION VI. HEALTH HAZARD DATA

**Route of Entry:**              X      Inhalation                      X      Skin                                  X      Ingestion

**Health Hazards (Acute and Chronic): Nitric Acid:** Nitric acid may be fatal if inhaled, swallowed, or absorbed through the skin. This material causes burns and is extremely destructive to tissue of the mucous membranes and upper respiratory tract, eyes, and skin. Inhalation may be fatal as a result of spasm, inflammation, and edema of the larynx and bronchi, chemical pneumonitis, and pulmonary edema. Symptoms of exposure may include burning sensation, coughing, wheezing, laryngitis, shortness of breath, headache, nausea, and vomiting.

**Medical Conditions Generally Aggravated by Exposure:** Eye disorders, respiratory disorders, skin disorders, and allergies.

**Listed as a Carcinogen/Potential Carcinogen:**

|  | Yes               | No       |
|--|-------------------|----------|
| In the National Toxicology Program (NTP) Report on Carcinogens       | <u>          </u> | <u>X</u> |
| In the International Agency for Research on Cancer (IARC) Monographs | <u>          </u> | <u>X</u> |
| By the Occupational Safety and Health Administration (OSHA)          | <u>          </u> | <u>X</u> |

**EMERGENCY AND FIRST AID PROCEDURES :**

**Skin Contact:** Remove contaminated shoes and clothing. Rinse affected area with large amounts of water followed by washing the area with soap and water. Watch for chemical irritations and treat them accordingly. Obtain medical assistance if necessary.

**Eye Contact:** Immediately flush eyes, including under the eyelids, with copious amounts of water for at least 15 minutes. Obtain medical assistance.

**Inhalation:** If inhaled, move the victim to fresh air. If breathing is difficult, give oxygen; if the victim is not breathing, give artificial respiration. Obtain medical assistance if necessary.

**Ingestion:** If ingestion occurs, wash out mouth with water. **DO NOT** induce vomiting. Obtain medical assistance immediately.

**NOTE (Nitric Acid):** Wash affected skin areas with 5 % solution of sodium bicarbonate ( $\text{NaHCO}_3$ ). If ingested, the risk versus the benefit of the passage of a naso-gastric tube is debatable. Activated charcoal is of no value. **DO NOT** give the exposed person bicarbonate to neutralize the material.

**TARGET ORGAN(S) OF ATTACK:** Skin, teeth, eyes, and upper respiratory tract.

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**SECTION VII. PRECAUTIONS FOR SAFE HANDLING AND USE**

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**Steps to be Taken in Case Material is Released or Spilled:** Notify safety personnel of spills. Spills should be handled according to radioactive spill procedures.

**Waste Disposal:** Follow all federal, state, and local laws governing disposal of radioactive materials.

**Handling and Storage:** The sample container should be handled by persons qualified to handle both radioactive material and strong acid solutions. Provide local exhaust or process enclosure ventilation system to ensure compliance with applicable exposure limits. Provide approved respiratory apparatus for non-routine or emergency use. Wear chemical resistant gloves and chemical safety glasses where contact with the liquid or high vapor concentrations may occur. An eye wash station and washing facilities should be readily available near handling and use areas.

**NOTE:** Contact lenses pose a special problem; soft lenses may absorb irritants and all lenses concentrate them. **DO NOT** wear contact lenses in the laboratory.

This material should be stored and used at a temperature between 5 °C and 65 °C.

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**SECTION VIII. SOURCE DATA/OTHER COMMENTS**

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**Sources:** MDL Information Systems, Inc., MSDS *Nitric Acid*, June 2, 1999.  
The Merck Index, 11th Ed., 1989.  
The Sigma-Aldrich Library of Chemical Safety Data, Ed. II, Vol. 2, 1988.

**Disclaimer:** Physical and chemical data contained in this MSDS are provided only for use in assessing the hazardous nature of the material. The MSDS was prepared carefully, using current references; however, NIST does not certify the data on the MSDS. The certified values for this material are given in the NIST Certificate of Analysis.